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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.005 Å R factor = 0.082 wR factor = 0.157 Data-to-parameter ratio = 11.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# N-(3-Fluorophenyl)-9H-xanthen-9-ylideneamine

In the title compound,  $C_{19}H_{12}FNO$ , the dihedral angle between the mean planes of the 9*H*-xanthene moiety and the 3-fluorophenyl group is 82.5 (1)°. An intramolecular C–  $H \cdots \pi$  interaction stabilizes the molecular conformation. Received 8 November 2004 Accepted 15 November 2004 Online 20 November 2004

## Comment

Molecular organization and molecular interactions are the features that are responsible for molecules exhibiting different functional properties. An understanding of non-covalent interactions becomes essential for interpreting and predicting relationships between chemical structure and function (Hunter et al., 2001). Crystal engineering, via manipulation of hydrogen bonding, has attracted a lot of interest in the recent literature (Aakeröy, 1997; Desiraju, 2000). Weak intermolecular forces of the type  $C-H \cdots \pi$  play an important role in various systems of biological and chemical interest (Nishio, 2004). Intramolecular C-H··· $\pi$  interactions are responsible for molecules adopting a particular conformation in the solid state (Jennings et al., 2001). A structural study of the title compound, (I), has been carried out as a case study where there is no possibility of formation of hydrogen bonds and hence it is thought to be suitable for the study of weak interactions.



A view of (I) with the atom-labelling scheme is shown in Fig. 1. The 9*H*-xanthene unit is almost planar, as indicated by the torsion angles (Table 1). The bond angle C12-N1-C14 of 126.7 (2)° is greater than the ideal bond angle value of 120° to avoid steric repulsion between atoms H15 and H1. Rotation of the 3-fluorophenyl group takes place and this is favoured because it leads to the formation of an intramolecular C-H $\cdots \pi$  interaction (Table 2) involving H1, which stabilizes the molecular conformation (Fig. 1).

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#### Figure 1

The molecular structure of (I), showing 40% probability ellipsoids. H atoms have been omitted for clarity, except for H1. The dashed line indicates the intramolecular  $C-H\cdots\pi$  interaction.

# **Experimental**

Compound (I) was synthesized according to a procedure reported in the literature (Nagarajan *et al.*, 1974). The compound was crystallized from a solution in ethyl acetate and hexane, by slow evaporation at *ca* 278 K.

Crystal data

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C<sub>19</sub>H<sub>12</sub>FNO

M_r = 289.30

Monoclinic, P_{2_1}/n

a = 14.175 (3) Å

b = 5.0634 (9) Å

c = 19.912 (3) Å

\beta = 102.794 (4)°

V = 1393.6 (4) Å<sup>3</sup>

Z = 4
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#### Data collection

Bruker SMART CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.902, T_{\max} = 0.949$ 10501 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.082$   $wR(F^2) = 0.157$  S = 1.122849 reflections 247 parameters All H-atom parameters refined MoKa radiation Cell parameters from 757 reflections  $\theta = 1.3-24.4^{\circ}$  $\mu = 0.10 \text{ mm}^{-1}$ T = 293 (2) KPrism, orange-yellow  $0.60 \times 0.60 \times 0.56 \text{ mm}$ 

 $D_{\rm x} = 1.379 {\rm Mg} {\rm m}^{-3}$ 

2849 independent reflections
1600 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.069$
$\theta_{\rm max} = 26.4^{\circ}$
$h = -17 \rightarrow 17$
$k = -6 \rightarrow 6$
$l = -24 \rightarrow 24$

$w = 1/[\sigma^2(F_o^2) + (0.0527P)^2]$
+ 0.0746P]
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.16 \text{ e} \text{ Å}^{-3}$
$\Delta \rho_{\rm min} = -0.17 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1			
Selected	geometric parameters	(Å,	°).

N1-C12	1.283 (3)	N1-C14	1.403 (4)
C12-N1-C14	126.7 (2)		
C14-N1-C12-C13 N1-C12-C11-C10 N1-C12-C13-C1	-6.1 (5) 1.4 (4) -5.8 (5)	C6-O1-C5-C13 C12-N1-C14-C15 C5-O1-C6-C11	1.7 (4) -82.4 (4) -0.8 (4)

# Table 2 Hydrogen-bonding geometry (Å, $^\circ).$

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D-\mathrm{H}\cdots A$		
$C1-H1\cdots Cg1$	0.98 (3)	2.76 (3)	3.630 (5)	147 (2)		
C-1 is the contracid of the C14 C10 ring						

Cg1 is the centroid of the C14–C19 ring.

All H atoms were located in difference Fourier maps and refined isotropically. The C–H bond distances are in the range 0.89 (3)– 1.08 (2) Å.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *PLATON* (Spek, 2003).

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