Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.005 Å R factor = 0.082 wR factor = 0.157 Data-to-parameter ratio = 11.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N-(3-Fluorophenyl)-9H-xanthen-9-ylideneamine

In the title compound, $C_{19}H_{12}FNO$, the dihedral angle between the mean planes of the 9*H*-xanthene moiety and the 3-fluorophenyl group is 82.5 (1)°. An intramolecular C– $H \cdots \pi$ interaction stabilizes the molecular conformation. Received 8 November 2004 Accepted 15 November 2004 Online 20 November 2004

Comment

Molecular organization and molecular interactions are the features that are responsible for molecules exhibiting different functional properties. An understanding of non-covalent interactions becomes essential for interpreting and predicting relationships between chemical structure and function (Hunter et al., 2001). Crystal engineering, via manipulation of hydrogen bonding, has attracted a lot of interest in the recent literature (Aakeröy, 1997; Desiraju, 2000). Weak intermolecular forces of the type $C-H \cdots \pi$ play an important role in various systems of biological and chemical interest (Nishio, 2004). Intramolecular C-H··· π interactions are responsible for molecules adopting a particular conformation in the solid state (Jennings et al., 2001). A structural study of the title compound, (I), has been carried out as a case study where there is no possibility of formation of hydrogen bonds and hence it is thought to be suitable for the study of weak interactions.



A view of (I) with the atom-labelling scheme is shown in Fig. 1. The 9*H*-xanthene unit is almost planar, as indicated by the torsion angles (Table 1). The bond angle C12-N1-C14 of 126.7 (2)° is greater than the ideal bond angle value of 120° to avoid steric repulsion between atoms H15 and H1. Rotation of the 3-fluorophenyl group takes place and this is favoured because it leads to the formation of an intramolecular C-H $\cdots \pi$ interaction (Table 2) involving H1, which stabilizes the molecular conformation (Fig. 1).

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Figure 1

The molecular structure of (I), showing 40% probability ellipsoids. H atoms have been omitted for clarity, except for H1. The dashed line indicates the intramolecular $C-H\cdots\pi$ interaction.

Experimental

Compound (I) was synthesized according to a procedure reported in the literature (Nagarajan *et al.*, 1974). The compound was crystallized from a solution in ethyl acetate and hexane, by slow evaporation at *ca* 278 K.

Crystal data

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C<sub>19</sub>H<sub>12</sub>FNO

M_r = 289.30

Monoclinic, P_{2_1}/n

a = 14.175 (3) Å

b = 5.0634 (9) Å

c = 19.912 (3) Å

\beta = 102.794 (4)°

V = 1393.6 (4) Å<sup>3</sup>

Z = 4
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Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.902, T_{\max} = 0.949$ 10501 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.082$ $wR(F^2) = 0.157$ S = 1.122849 reflections 247 parameters All H-atom parameters refined MoKa radiation Cell parameters from 757 reflections $\theta = 1.3-24.4^{\circ}$ $\mu = 0.10 \text{ mm}^{-1}$ T = 293 (2) KPrism, orange-yellow $0.60 \times 0.60 \times 0.56 \text{ mm}$

 $D_{\rm x} = 1.379 {\rm Mg} {\rm m}^{-3}$

2849 independent reflections
1600 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.069$
$\theta_{\rm max} = 26.4^{\circ}$
$h = -17 \rightarrow 17$
$k = -6 \rightarrow 6$
$l = -24 \rightarrow 24$

$w = 1/[\sigma^2(F_o^2) + (0.0527P)^2]$
+ 0.0746P]
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.16 \text{ e } \text{\AA}^{-3}$
$\Delta \rho_{\rm min} = -0.17 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1			
Selected	geometric parameters	(Å,	°).

N1-C12	1.283 (3)	N1-C14	1.403 (4)
C12-N1-C14	126.7 (2)		
C14-N1-C12-C13 N1-C12-C11-C10 N1-C12-C13-C1	-6.1 (5) 1.4 (4) -5.8 (5)	C6-O1-C5-C13 C12-N1-C14-C15 C5-O1-C6-C11	1.7 (4) -82.4 (4) -0.8 (4)

Table 2 Hydrogen-bonding geometry (Å, $^\circ).$

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D-\mathrm{H}\cdots A$		
$C1-H1\cdots Cg1$	0.98 (3)	2.76 (3)	3.630 (5)	147 (2)		
C-1 is the contracid of the C14 C10 ring						

Cg1 is the centroid of the C14–C19 ring.

All H atoms were located in difference Fourier maps and refined isotropically. The C–H bond distances are in the range 0.89 (3)– 1.08 (2) Å.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *PLATON* (Spek, 2003).

We thank the Department of Science and Technology, India, for data collection on the CCD facility set up under the IRFA–DST program. D. Chopra thanks CSIR, India, for a Junior Research Fellowship.

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